

# Math Station 1

$$p^2 + 2pq + q^2 = 1$$

$$p + q = 1$$

## HARDY-WEINBERG EQUATIONS

$p$  = frequency of the dominant allele in a population

$q$  = frequency of the recessive allele in a population

A population of beavers revealed that 12 of them show a rare recessive condition. The other 88 beavers in this population show no condition.

If this population is in Hardy Weinberg equilibrium, what is the frequency of the dominant allele?

Give your answer to the nearest hundredth

# Math Station 2

$$p^2 + 2pq + q^2 = 1$$

$$p + q = 1$$

## HARDY-WEINBERG EQUATIONS

$p$  = frequency of the dominant allele in a population

$q$  = frequency of the recessive allele in a population

A population of cacti revealed that 21% have a dominant condition.

If this population is in Hardy Weinberg equilibrium, what is the frequency of the heterozygous?

Give your answer to the nearest tenths place

# Math Station 3

40 fruit flies were put in a choice chamber. The results of their location are shown below:

Time	Number of fruit flies on orange	Number of fruit flies on apple
0 min	20	20
10 min	8	32

Determine the chi square value after 10 min.

Round to the nearest whole number.

Chi-Square

$$\chi^2 = \sum \frac{(O - E)^2}{E}$$

CHI-SQUARE TABLE

Degrees of Freedom

p	1	2	3	4	5	6	7	8
0.05	3.84	5.99	7.82	9.49	11.07	12.59	14.07	15.51
0.01	6.64	9.32	11.34	13.28	15.09	16.81	18.48	20.09

# Math Station 4

$$\chi^2 = \sum \frac{(O - E)^2}{E}$$

CHI-SQUARE TABLE								
Degrees of Freedom								
p	1	2	3	4	5	6	7	8
0.05	3.84	5.99	7.82	9.49	11.07	12.59	14.07	15.51
0.01	6.64	9.32	11.34	13.28	15.09	16.81	18.48	20.09

In pea plants, purple (P) is dominant to white (p). One hundred purple pea plants were crossed in the P generation. The results of the F1 are shown below:

Phenotype	Number observed in F1
Purple	63
White	37

Determine the chi square value.

Round to the nearest tenths place.

# Math Station 5

A sample of lizards in the Caribbean show variation in their back-leg length.

Given the following data, determine the standard error for this data.

**Length(cm): 2.0, 2.2, 3.2, and 2.5**

Calculate the standard error.

Round the answer to the nearest hundredth.

STATISTICAL
Standard Error
$SE_{\bar{y}} = \frac{s}{\sqrt{n}}$
Standard Deviation
$s = \sqrt{\frac{\sum(x_i - \bar{x})^2}{n-1}}$

# Math Station 6

RATE AND GROWTH	
Rate	$dY/dt$ = amount of change
$dY/dt$	$t$ = time
Population Growth	$B$ = birth rate
$dN/dt = B - D$	$D$ = death rate
Exponential Growth	$N$ = population size
$\frac{dN}{dt} = r_{max} N$	$K$ = carrying capacity
Logistic Growth	$r_{max}$ = maximum per capita growth rate of population
$\frac{dN}{dt} = r_{max} N \left( \frac{K - N}{K} \right)$	

A population of Spotted Fritillary butterflies has 200 individuals and this population exhibits logistic growth. If the carrying capacity is 500 butterflies and  $r = 0.1$  individuals/(individuals x month), what is the maximum population growth rate for the population?

Round your answer to the nearest tenths.

# Math Station 7

The molar concentration of a sugar solution in an open beaker has been determined to be 0.3M. Calculate the solute potential at 27 degrees Celsius.

Round your answer to the nearest tenths.

## Water Potential ( $\Psi$ )

$$\Psi = \Psi_p + \Psi_s$$

$\Psi_p$  = pressure potential

$\Psi_s$  = solute potential

The water potential will be equal to the solute potential of a solution in an open container, since the pressure potential of the solution in an open container is zero.

## The Solute Potential of the Solution

$$\Psi_s = -iCRT$$

$i$  = ionization constant (For sucrose this is 1.0 because sucrose does not ionize in water.)

$C$  = molar concentration

$R$  = pressure constant ( $R = 0.0831$  liter bars/mole K)

$T$  = temperature in Kelvin ( $273 + ^\circ\text{C}$ )

# Math Station 8

<p><b>Temperature Coefficient <math>Q_{10}</math></b></p> $Q_{10} = \left(\frac{k_2}{k_1}\right)^{\frac{10}{t_2 - t_1}}$ <p><b>Primary Productivity Calculation</b></p> <p>mg O<sub>2</sub>/L x 0.698 = mL O<sub>2</sub>/L</p> <p>mL O<sub>2</sub>/L x 0.536 = mg carbon fixed/L</p>	<p><math>t_2</math> = higher temperature</p> <p><math>t_1</math> = lower temperature</p> <p><math>k_2</math> = metabolic rate at <math>t_2</math></p> <p><math>k_1</math> = metabolic rate at <math>t_1</math></p> <p><math>Q_{10}</math> = the <i>factor</i> by which the reaction rate increases when the temperature is raised by ten degrees</p>
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Data taken to determine the effect of temperature on the rate of respiration in a lizard is given in the table below.

Temperature (C)	Heart Rate
15	60
25	100

Calculate  $Q_{10}$  for this data.

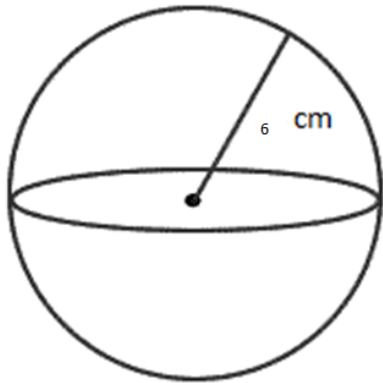
Round to the nearest whole number.



## AP Biology Math Stations

# Math Station 9

SURFACE AREA AND VOLUME	
Volume of a Sphere $V = \frac{4}{3} \pi r^3$	$r$ = radius
Volume of a Cube (or Square Column) $V = l w h$	$l$ = length
Volume of a Column $V = \pi r^2 h$	$h$ = height
Surface Area of a Sphere $A = 4 \pi r^2$	$w$ = width
Surface Area of a Cube $A = 6 a$	$A$ = surface area
Surface Area of a Rectangular Solid $A = \Sigma$ (surface area of each side)	$V$ = volume
	$\Sigma$ = Sum of all
	$a$ = surface area of one side of the cube



What is the Surface Area/Volume ratio for this cell?

Round your answer to the nearest hundredths.

Dilution – used to create a dilute solution from a concentrated stock solution

$$C_i V_i = C_f V_f$$

i = initial (starting)

C = concentration of solute

f = final (desired)

V = volume of solution

# Math Station 10

A student has a 2 g/L solution. He dilutes it and creates 3 L of a 1 g/L solution.

How much volume of the original solution did he dilute?

Round to the nearest tenths.

## AP Biology Math Stations

$$^{\circ}\text{C} = (^{\circ}\text{F} - 32) * 5/9$$

$$^{\circ}\text{F} = ^{\circ}\text{C} * 9/5 + 32$$

$$\text{K} = ^{\circ}\text{C} + 273$$

### Gibbs Free Energy

$$\Delta G = \Delta H - T\Delta S$$

$\Delta G$  = change in Gibbs free energy

$\Delta S$  = change in entropy

$\Delta H$  = change in enthalpy

$T$  = absolute temperature (in Kelvin)

$$\text{pH} = -\log [\text{H}^+]$$

# Math Station 11

Calculate the Gibbs free energy change ( $\Delta G$ ) for the following chemical reaction:  $\text{ATP} \rightarrow \text{ADP} + \text{P}_i$

The reaction occurs at 68  $^{\circ}\text{F}$ , the change in heat ( $\Delta H$ ) = 19,070 cal, and the change in entropy ( $\Delta S$ ) = 90 cal/K.

Determine the **kcal** value.

Round to the nearest tenths place.

## AP Biology Math Stations

# Math Station 12

SURFACE AREA AND VOLUME	
Volume of a Sphere $V = 4/3 \pi r^3$	$r$ = radius
Volume of a Cube (or Square Column) $V = l w h$	$l$ = length
Volume of a Column $V = \pi r^2 h$	$h$ = height
Surface Area of a Sphere $A = 4 \pi r^2$	$w$ = width
Surface Area of a Cube $A = 6 a$	$A$ = surface area
Surface Area of a Rectangular Solid $A = \Sigma$ (surface area of each side)	$V$ = volume
	$\Sigma$ = Sum of all
	$a$ = surface area of one side of the cube

A student has cut two blocks of phenolphthalein agar to measure the rate of diffusion in a vinegar solution.

Cube A: 4cm on each side

Cube B: 2 cm on each side

What the difference in surface area between the two blocks?  
Round answer to the nearest tenths place.

## AP Biology Math Stations

During AP Review Week, I am doing stations. One station is a concept map for the Big Idea, I assign group. Another is taking practice exam together. Another is reviewing Bozeman videos. The last are these math stations